$\qquad$ Date $\qquad$ Class $\qquad$

## Lesson 1

atom small particle that is a building block of matter
compound substance containing atoms of two or more different elements chemically bonded together
dissolve to form a solution by mixing evenly
element substance that consists of just one type of atom
heterogeneous mixture type of mixture in which the individual substances are not evenly mixed
homogeneous mixture type of mixture in which the individual substances are evenly mixed
matter anything that has mass and
takes up space
mixture matter that can vary in composition
substance matter with a composition that is always the same
unique having nothing else like it

## Lesson 2

bond force between atoms or groups of atoms
density mass per unit volume of a substance
mass amount of matter in an object
physical property characteristic of matter that you can observe or measure without changing the identity of the matter
property characteristic used to describe something
solubility ability of one substance to dissolve in another
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## 

## Lesson 3

physical change change in size, shape, form, or state of matter in which the matter's identity stays the same

## Lesson 4

chemical change change in matter in which the substances that make up the matter change into other substances with new physical and chemical properties
chemical property characteristic of matter that can be observed as it changes to a different type of matter
concentration amount of substance in a certain volume

